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Diaqua[(*E*)-2-(2-oxidobenzylideneamino)-2-phenylacetato]zinc(II) dimethyl sulfoxide monosolvate

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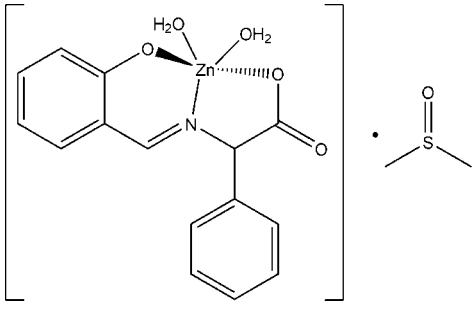
Key indicators: single-crystal X-ray study; T = 291 K; mean σ(C–C) = 0.004 Å; R factor = 0.033; wR factor = 0.098; data-to-parameter ratio = 18.3.

In the title compound, $[Zn(C_{15}H_{11}NO_3)(H_2O)_2]\cdot C_2H_6OS$, the Zn(II) ion is coordinated by two O atoms and one N atom of the deprotonated chelate ligand and two water molecules in a distorted trigonal bipyramidal coordination environment. A linear supramolecular structure built from O—H···O hydrogen bonds runs parallel to [100].

Related literature

For the synthesis of (*E*)-2-(2-hydroxybenzylideneamino)-2-phenylacetic acid, see Audriceth *et al.* (1954). For a related zinc complex, see: You *et al.* (2008).



Experimental

Crystal data

$[Zn(C_{15}H_{11}NO_3)(H_2O)_2]\cdot C_2H_6OS$
M_r = 432.78
Triclinic, $P\overline{1}$
a = 7.331 (4) Å
b = 9.318 (5) Å
c = 14.578 (9) Å
α = 81.91 (2)°
β = 81.37 (2)°
γ = 80.18 (2)°
V = 963.4 (9) Å^3
Z = 2
Mo $K\alpha$ radiation
μ = 1.42 mm^{-1}
T = 291 K
0.19 × 0.15 × 0.13 mm

Data collection

Rigaku R-AXIS RAPID diffractometer
Absorption correction: multi-scan (*ABSCOR*; Higashi, 1995)
T_{min} = 0.777, T_{max} = 0.837
9415 measured reflections
4331 independent reflections
3751 reflections with $I > 2\sigma(I)$
R_{int} = 0.024

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.033
$wR(F^2)$ = 0.098
S = 1.15
4331 reflections
237 parameters
H-atom parameters constrained
$\Delta\rho_{max}$ = 0.48 e Å^{-3}
$\Delta\rho_{min}$ = −0.28 e Å^{-3}

Table 1
Selected geometric parameters (Å, °).

N1—Zn1	2.0305 (19)	O4—Zn1	2.009 (2)
O1—Zn1	2.1047 (17)	O5—Zn1	2.0015 (19)
O3—Zn1	1.9742 (18)		

Table 2
Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
O4—H23···O6	0.85	1.90	2.725 (3)	164
O4—H24···O2^{i}	0.85	1.88	2.695 (3)	160
O5—H21···O2^{ii}	0.85	1.81	2.629 (2)	161
O5—H22···O6^{iii}	0.85	1.93	2.742 (3)	159

Symmetry codes: (i) $-x+2, -y+1, -z$; (ii) $x-1, y, z$; (iii) $-x+1, -y+1, -z$.

Data collection: *RAPID-AUTO* (Rigaku, 1998); cell refinement: *RAPID-AUTO*; data reduction: *CrystalStructure* (Rigaku/MSC, 2002); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXL97*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: NG2591).

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supplementary materials

Acta Cryst. (2009). E**65**, m734 [**doi:10.1107/S1600536809020741**]

Diaqua[(*E*)-2-(2-oxidobenzylideneamino)-2-phenylacetato]zinc(II) dimethyl sulfoxide monosolvate

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Comment

The continuous interest in designing and making novel Schiff base ligand and transition-metal complexes have persisted because of their impressive catalytic property. Recently, our group have reported a Schiff base and Zn(II) complex (You *et al.*, 2008). As the continually working, we report a new title Schiff base complex, herein, synthesized by the reaction of (*E*)-2-(2-hydroxybenzylideneamino)-2-phenylacetic acid and $Zn(OAc)_2$.

As shown in Fig. 1, Zn^{II} ion is five-coordinate in a slightly distorted trigonal-bipyramidal coordination environment, two water molecules and one N formed the equatorial plane and two deprotonated O atoms take up the apices positions.

The cocrystalized dimethylsulfoxide molecules link the discrete coordinate compound to a one-dimensional tubal supramolecular structure, *via* the O—H···O hydrogen bonds parallel to [100] (Fig. 2).

Experimental

(*E*)-2-(2-hydroxybenzylideneamino)-2-phenylacetic acid was prepared of 2-amino-2-phenylacetic acid and 2-hydroxybenzaldehyde in aqueous solution (Audriceth *et al.*, 1954). (*E*)-2-(2-hydroxybenzylideneamino)-2-phenylacetic acid (0.255 g, 1 mmol) and $Zn(OAc)_2$ (0.190 g, 1 mmol) dissolved in hot aqueous solution (20 ml) then refluxed for 1 huor. After cooling to room temperature the solution was filtered, the residue was recrystaled in DMSO and methanol (10/1, V/V) solution, several days latter, a suitable for X-ray diffraction yellow crystal was obtained.

Refinement

H atoms bound to C atoms were placed in calculated positions and treated as riding on their parent atoms, with C—H = 0.93 Å (aromatic C), C—H = 0.98 Å (methylene C), C—H = 0.96 Å (methyl C) and with U_{iso}(H) = 1.2Ueq(C). Water H atoms were initially located in a difference Fourier map, but they were treated as riding on their parent atoms with O—H = 0.85 Å and with with U_{iso}(H) = 1.5U_{eq}(O).

Figures

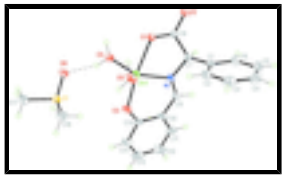

Fig. 1. The molecular structure of (I), showing displacement ellipsoids at the 30% probability level for non-H atoms.

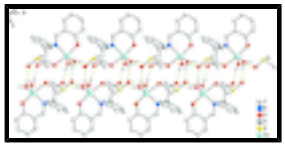

Fig. 2. A partial packing view, showing the one-dimensional tubal supramolecular structure. H atoms not involved in hydrogen bonds have been omitted for clarity.

Diaqua[(*E*)-2-(2-oxidobenzylideneamino)-2-phenylacetato]zinc(II) dimethyl sulfoxide solvate

Crystal data

$[Zn(C_{15}H_{11}NO_3)(H_2O)_2]\cdot C_2H_6OS$	$Z = 2$
$M_r = 432.78$	$F_{000} = 448$
Triclinic, $P\overline{1}$	$D_x = 1.492$ Mg m^{-3}
Hall symbol: -P 1	Mo $K\alpha$ radiation $\lambda = 0.71073$ Å
$a = 7.331$ (4) Å	Cell parameters from 8052 reflections
$b = 9.318$ (5) Å	$\theta = 3.0$–27.5º
$c = 14.578$ (9) Å	$\mu = 1.42$ mm^{-1}
$\alpha = 81.91$ (2)º	$T = 291$ K
$\beta = 81.37$ (2)º	Block, colorless
$\gamma = 80.18$ (2)º	0.19 × 0.15 × 0.13 mm
$V = 963.4$ (9) Å^3	

Data collection

Rigaku R-AXIS RAPID diffractometer	4331 independent reflections
Radiation source: fine-focus sealed tube	3751 reflections with $I > 2\sigma(I)$
Monochromator: graphite	$R_{int} = 0.024$
$T = 291$ K	$\theta_{max} = 27.5$º
ω scan	$\theta_{min} = 3.0$º
Absorption correction: multi-scan (ABSCOR; Higashi, 1995)	$h = -9\rightarrow8$
$T_{min} = 0.777$, $T_{max} = 0.837$	$k = -12\rightarrow12$
9415 measured reflections	$l = -18\rightarrow18$

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.033$	H-atom parameters constrained
$wR(F^2) = 0.098$	$w = 1/[\sigma^2(F_o^2) + (0.0512P)^2 + 0.2263P]$ where $P = (F_o^2 + 2F_c^2)/3$
$S = 1.15$	$(\Delta/\sigma)_{max} = 0.001$
4331 reflections	$\Delta\rho_{max} = 0.48$ e Å^{-3}
237 parameters	$\Delta\rho_{min} = -0.28$ e Å^{-3}

Primary atom site location: structure-invariant direct methods

Extinction correction: none

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted *R*-factor *wR* and goodness of fit *S* are based on F^2, conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative F^2. The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating *R*-factors(gt) *etc*. and is not relevant to the choice of reflections for refinement. *R*-factors based on F^2 are statistically about twice as large as those based on *F*, and *R*- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	*y*	*z*	U_{iso}*/U_{eq}
C1	0.9379 (3)	0.7675 (2)	0.26445 (14)	0.0316 (4)
C2	0.7789 (4)	0.8689 (3)	0.24967 (18)	0.0451 (5)
H2	0.6891	0.8462	0.2176	0.054*
C3	0.7553 (5)	1.0041 (3)	0.2831 (2)	0.0589 (7)
H3	0.6491	1.0719	0.2736	0.071*
C4	0.8891 (5)	1.0382 (3)	0.3305 (2)	0.0671 (9)
H4	0.8718	1.1284	0.3535	0.080*
C5	1.0461 (5)	0.9402 (3)	0.3436 (2)	0.0612 (8)
H5	1.1362	0.9643	0.3749	0.073*
C6	1.0726 (4)	0.8041 (3)	0.31051 (17)	0.0457 (6)
H6	1.1806	0.7379	0.3193	0.055*
C7	0.9648 (3)	0.6175 (2)	0.23069 (14)	0.0286 (4)
H7	1.0767	0.5582	0.2531	0.034*
C8	0.9913 (3)	0.6342 (2)	0.12286 (14)	0.0287 (4)
C9	0.7835 (3)	0.4801 (2)	0.34886 (15)	0.0339 (4)
H9	0.8721	0.4896	0.3862	0.041*
C10	0.6363 (3)	0.3983 (2)	0.39195 (15)	0.0356 (5)
C11	0.6505 (4)	0.3309 (3)	0.48364 (17)	0.0521 (6)
H11	0.7511	0.3425	0.5124	0.063*
C12	0.5219 (5)	0.2490 (4)	0.5320 (2)	0.0635 (8)
H12	0.5353	0.2044	0.5923	0.076*
C13	0.3711 (5)	0.2339 (4)	0.4894 (2)	0.0655 (8)
H13	0.2828	0.1781	0.5217	0.079*
C14	0.3488 (4)	0.2995 (3)	0.4004 (2)	0.0559 (7)
H14	0.2444	0.2886	0.3743	0.067*
C15	0.4808 (3)	0.3834 (3)	0.34729 (15)	0.0392 (5)
C16	0.1636 (4)	0.2801 (3)	0.1543 (2)	0.0527 (6)
H16A	0.2255	0.3352	0.1877	0.079*
H16B	0.0506	0.2573	0.1919	0.079*
H16C	0.1341	0.3372	0.0969	0.079*
C17	0.1686 (4)	0.0495 (3)	0.0617 (2)	0.0495 (6)

H17A	0.1290	0.1261	0.0143	0.074*
H17B	0.0612	0.0201	0.1014	0.074*
H17C	0.2381	−0.0330	0.0327	0.074*
N1	0.8040 (2)	0.54080 (19)	0.26458 (12)	0.0301 (4)
O1	0.8636 (2)	0.60996 (18)	0.08213 (10)	0.0366 (3)
O2	1.1404 (2)	0.67381 (19)	0.08264 (11)	0.0400 (4)
O3	0.4514 (2)	0.4435 (2)	0.26298 (12)	0.0505 (4)
O4	0.6986 (3)	0.3510 (2)	0.09429 (13)	0.0581 (5)
H23	0.6190	0.2942	0.0951	0.087*
H24	0.7571	0.3634	0.0395	0.087*
O5	0.4726 (2)	0.7027 (2)	0.11810 (12)	0.0471 (4)
H21	0.3578	0.6969	0.1193	0.071*
H22	0.5174	0.7309	0.0626	0.071*
O6	0.4740 (2)	0.16387 (19)	0.06180 (12)	0.0447 (4)
S1	0.31254 (8)	0.11488 (6)	0.12970 (4)	0.03825 (14)
Zn1	0.63958 (3)	0.52441 (3)	0.167315 (16)	0.03534 (10)

Atomic displacement parameters (Å^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0338 (10)	0.0317 (10)	0.0280 (9)	−0.0083 (8)	0.0023 (8)	−0.0019 (8)
C2	0.0419 (13)	0.0445 (13)	0.0463 (13)	−0.0040 (11)	−0.0015 (10)	−0.0040 (11)
C3	0.0648 (18)	0.0380 (13)	0.0642 (18)	0.0045 (13)	0.0099 (14)	−0.0074 (13)
C4	0.104 (3)	0.0411 (15)	0.0562 (17)	−0.0236 (17)	0.0161 (17)	−0.0183 (13)
C5	0.080 (2)	0.0554 (17)	0.0589 (17)	−0.0311 (16)	−0.0067 (15)	−0.0191 (14)
C6	0.0504 (14)	0.0492 (14)	0.0420 (13)	−0.0155 (12)	−0.0069 (11)	−0.0102 (11)
C7	0.0240 (9)	0.0327 (10)	0.0301 (10)	−0.0082 (8)	−0.0031 (7)	−0.0031 (8)
C8	0.0257 (9)	0.0303 (10)	0.0296 (9)	−0.0069 (8)	0.0021 (7)	−0.0040 (8)
C9	0.0360 (11)	0.0367 (11)	0.0287 (10)	−0.0072 (9)	−0.0026 (8)	−0.0023 (8)
C10	0.0412 (12)	0.0338 (11)	0.0285 (10)	−0.0080 (9)	0.0059 (9)	−0.0009 (8)
C11	0.0544 (15)	0.0632 (16)	0.0338 (12)	−0.0107 (13)	0.0012 (11)	0.0063 (11)
C12	0.070 (2)	0.071 (2)	0.0398 (14)	−0.0145 (16)	0.0081 (13)	0.0177 (13)
C13	0.073 (2)	0.0637 (18)	0.0534 (16)	−0.0298 (16)	0.0221 (15)	0.0079 (14)
C14	0.0502 (15)	0.0702 (18)	0.0470 (14)	−0.0287 (14)	0.0107 (12)	−0.0001 (13)
C15	0.0409 (12)	0.0429 (12)	0.0321 (11)	−0.0141 (10)	0.0088 (9)	−0.0035 (9)
C16	0.0515 (15)	0.0485 (14)	0.0578 (16)	−0.0065 (12)	0.0028 (12)	−0.0166 (12)
C17	0.0400 (13)	0.0417 (13)	0.0701 (17)	−0.0152 (11)	−0.0069 (12)	−0.0076 (12)
N1	0.0288 (8)	0.0341 (9)	0.0271 (8)	−0.0091 (7)	−0.0001 (7)	−0.0014 (7)
O1	0.0290 (7)	0.0560 (10)	0.0274 (7)	−0.0179 (7)	−0.0024 (6)	−0.0014 (7)
O2	0.0281 (7)	0.0587 (10)	0.0355 (8)	−0.0200 (7)	0.0042 (6)	−0.0064 (7)
O3	0.0414 (9)	0.0763 (12)	0.0367 (9)	−0.0303 (9)	−0.0009 (7)	0.0050 (8)
O4	0.0618 (12)	0.0754 (13)	0.0467 (10)	−0.0448 (10)	0.0172 (9)	−0.0239 (9)
O5	0.0268 (8)	0.0655 (11)	0.0470 (9)	−0.0145 (8)	−0.0050 (7)	0.0094 (8)
O6	0.0350 (8)	0.0511 (10)	0.0497 (10)	−0.0179 (7)	0.0021 (7)	−0.0060 (8)
S1	0.0338 (3)	0.0366 (3)	0.0427 (3)	−0.0104 (2)	−0.0032 (2)	0.0048 (2)
Zn1	0.02999 (15)	0.05017 (18)	0.02839 (14)	−0.01813 (11)	−0.00285 (9)	0.00044 (10)

Geometric parameters (Å, °)

C1—C6	1.386 (3)	C12—H12	0.9300
C1—C2	1.393 (3)	C13—C14	1.376 (4)
C1—C7	1.519 (3)	C13—H13	0.9300
C2—C3	1.388 (4)	C14—C15	1.418 (3)
C2—H2	0.9300	C14—H14	0.9300
C3—C4	1.383 (5)	C15—O3	1.310 (3)
C3—H3	0.9300	C16—S1	1.777 (3)
C4—C5	1.361 (5)	C16—H16A	0.9600
C4—H4	0.9300	C16—H16B	0.9600
C5—C6	1.393 (4)	C16—H16C	0.9600
C5—H5	0.9300	C17—S1	1.780 (3)
C6—H6	0.9300	C17—H17A	0.9600
C7—N1	1.469 (3)	C17—H17B	0.9600
C7—C8	1.544 (3)	C17—H17C	0.9600
C7—H7	0.9800	N1—Zn1	2.0305 (19)
C8—O1	1.248 (2)	O1—Zn1	2.1047 (17)
C8—O2	1.248 (2)	O3—Zn1	1.9742 (18)
C9—N1	1.276 (3)	O4—Zn1	2.009 (2)
C9—C10	1.443 (3)	O4—H23	0.8500
C9—H9	0.9300	O4—H24	0.8500
C10—C11	1.405 (3)	O5—Zn1	2.0015 (19)
C10—C15	1.430 (3)	O5—H21	0.8500
C11—C12	1.365 (4)	O5—H22	0.8499
C11—H11	0.9300	O6—S1	1.5147 (18)
C12—C13	1.382 (5)		
C6—C1—C2	119.6 (2)	C13—C14—C15	121.7 (3)
C6—C1—C7	119.8 (2)	C13—C14—H14	119.2
C2—C1—C7	120.64 (19)	C15—C14—H14	119.2
C3—C2—C1	119.6 (3)	O3—C15—C14	118.8 (2)
C3—C2—H2	120.2	O3—C15—C10	124.8 (2)
C1—C2—H2	120.2	C14—C15—C10	116.4 (2)
C4—C3—C2	120.3 (3)	S1—C16—H16A	109.5
C4—C3—H3	119.9	S1—C16—H16B	109.5
C2—C3—H3	119.9	H16A—C16—H16B	109.5
C5—C4—C3	120.2 (2)	S1—C16—H16C	109.5
C5—C4—H4	119.9	H16A—C16—H16C	109.5
C3—C4—H4	119.9	H16B—C16—H16C	109.5
C4—C5—C6	120.4 (3)	S1—C17—H17A	109.5
C4—C5—H5	119.8	S1—C17—H17B	109.5
C6—C5—H5	119.8	H17A—C17—H17B	109.5
C1—C6—C5	119.9 (3)	S1—C17—H17C	109.5
C1—C6—H6	120.1	H17A—C17—H17C	109.5
C5—C6—H6	120.1	H17B—C17—H17C	109.5
N1—C7—C1	111.97 (16)	C9—N1—C7	119.69 (18)
N1—C7—C8	108.26 (15)	C9—N1—Zn1	124.24 (15)
C1—C7—C8	109.72 (17)	C7—N1—Zn1	115.85 (13)

N1—C7—H7	108.9	C8—O1—Zn1	116.45 (13)
C1—C7—H7	108.9	C15—O3—Zn1	125.28 (15)
C8—C7—H7	108.9	Zn1—O4—H23	121.1
O1—C8—O2	124.70 (19)	Zn1—O4—H24	117.3
O1—C8—C7	118.68 (17)	H23—O4—H24	109.4
O2—C8—C7	116.62 (18)	Zn1—O5—H21	118.0
N1—C9—C10	126.4 (2)	Zn1—O5—H22	108.9
N1—C9—H9	116.8	H21—O5—H22	109.0
C10—C9—H9	116.8	O6—S1—C16	104.74 (13)
C11—C10—C15	119.5 (2)	O6—S1—C17	106.13 (13)
C11—C10—C9	116.5 (2)	C16—S1—C17	98.10 (13)
C15—C10—C9	124.1 (2)	O3—Zn1—O5	97.31 (8)
C12—C11—C10	122.4 (3)	O3—Zn1—O4	96.26 (9)
C12—C11—H11	118.8	O5—Zn1—O4	118.61 (9)
C10—C11—H11	118.8	O3—Zn1—N1	92.18 (8)
C11—C12—C13	118.6 (3)	O5—Zn1—N1	119.53 (8)
C11—C12—H12	120.7	O4—Zn1—N1	119.38 (9)
C13—C12—H12	120.7	O3—Zn1—O1	171.33 (6)
C14—C13—C12	121.5 (3)	O5—Zn1—O1	87.49 (8)
C14—C13—H13	119.3	O4—Zn1—O1	87.69 (7)
C12—C13—H13	119.3	N1—Zn1—O1	79.16 (7)
C6—C1—C2—C3	−1.5 (4)	C11—C10—C15—C14	0.0 (3)
C7—C1—C2—C3	178.5 (2)	C9—C10—C15—C14	−179.7 (2)
C1—C2—C3—C4	0.2 (4)	C10—C9—N1—C7	−178.51 (19)
C2—C3—C4—C5	0.9 (5)	C10—C9—N1—Zn1	−4.2 (3)
C3—C4—C5—C6	−0.8 (5)	C1—C7—N1—C9	−77.4 (2)
C2—C1—C6—C5	1.6 (4)	C8—C7—N1—C9	161.56 (19)
C7—C1—C6—C5	−178.4 (2)	C1—C7—N1—Zn1	107.87 (16)
C4—C5—C6—C1	−0.5 (4)	C8—C7—N1—Zn1	−13.2 (2)
C6—C1—C7—N1	126.6 (2)	O2—C8—O1—Zn1	174.12 (17)
C2—C1—C7—N1	−53.4 (3)	C7—C8—O1—Zn1	−7.0 (2)
C6—C1—C7—C8	−113.2 (2)	C14—C15—O3—Zn1	−165.10 (19)
C2—C1—C7—C8	66.9 (3)	C10—C15—O3—Zn1	16.3 (3)
N1—C7—C8—O1	13.2 (3)	C15—O3—Zn1—O5	−139.4 (2)
C1—C7—C8—O1	−109.3 (2)	C15—O3—Zn1—O4	100.7 (2)
N1—C7—C8—O2	−167.87 (18)	C15—O3—Zn1—N1	−19.2 (2)
C1—C7—C8—O2	69.7 (2)	C9—N1—Zn1—O3	13.31 (19)
N1—C9—C10—C11	174.9 (2)	C7—N1—Zn1—O3	−172.18 (14)
N1—C9—C10—C15	−5.4 (4)	C9—N1—Zn1—O5	112.98 (18)
C15—C10—C11—C12	1.0 (4)	C7—N1—Zn1—O5	−72.52 (15)
C9—C10—C11—C12	−179.3 (3)	C9—N1—Zn1—O4	−85.13 (19)
C10—C11—C12—C13	−0.9 (5)	C7—N1—Zn1—O4	89.38 (15)
C11—C12—C13—C14	−0.2 (5)	C9—N1—Zn1—O1	−166.21 (19)
C12—C13—C14—C15	1.2 (5)	C7—N1—Zn1—O1	8.29 (13)
C13—C14—C15—O3	−179.8 (3)	C8—O1—Zn1—O5	120.12 (16)
C13—C14—C15—C10	−1.1 (4)	C8—O1—Zn1—O4	−121.10 (16)
C11—C10—C15—O3	178.6 (2)	C8—O1—Zn1—N1	−0.59 (15)
C9—C10—C15—O3	−1.1 (4)		

Hydrogen-bond geometry (Å, °)

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
O4—H23···O6	0.85	1.90	2.725 (3)	164
O4—H24···O2^{i}	0.85	1.88	2.695 (3)	160
O5—H21···O2^{ii}	0.85	1.81	2.629 (2)	161
O5—H22···O6^{iii}	0.85	1.93	2.742 (3)	159

Symmetry codes: (i) −*x*+2, −*y*+1, −*z*; (ii) *x*−1, *y*, *z*; (iii) −*x*+1, −*y*+1, −*z*.

Fig. 1

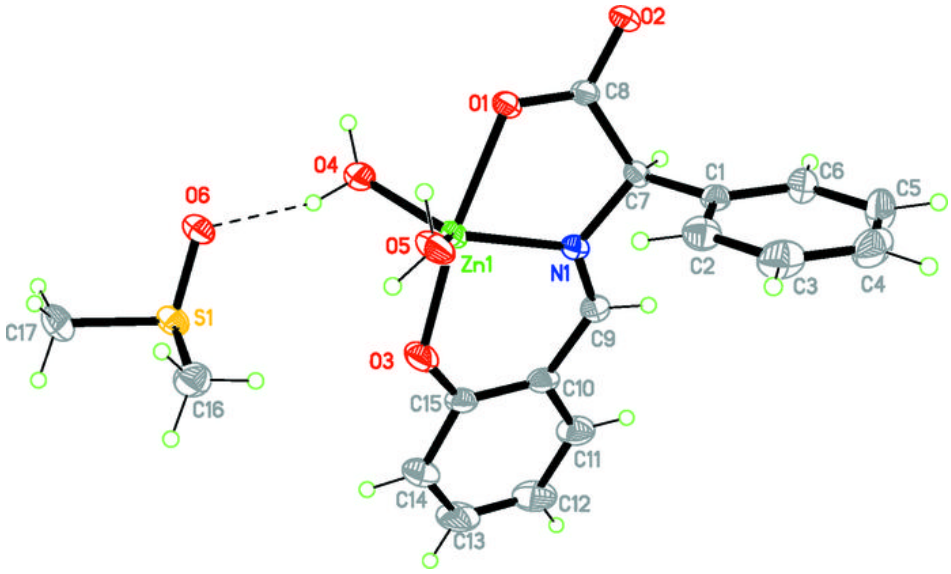
O2
O1
C8
O4
C1
C6
C5
O6
C7
O5
Zn1
N1
C2
C3
C4
S1
C17
C9
O3
C10
C15
C16
C11
C14
C12
C13

Fig. 2